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# Material Safety Data Sheet Carbon Tetrabromide MSDS

### Section 1: Chemical Product and Company Identification

Product Name: Carbon Tetrabromide

Catalog Codes: 10456

CAS#: 558-13-4

RTECS: FG4275000

TSCA: TSCA 8(b) inventory: Carbon Tetrabromide

Cl#: Not available.

**Synonym:** Tetrabromomethane; Methane, tetrabromide; Carbon Bromide

Chemical Name: Carbon Tetrabromide

Chemical Formula: C-Br4

### Contact Information:

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# Section 2: Composition and Information on Ingredients

#### **Composition:**

Name	CAS #	% by Weight
Carbon Tetrabromide	558-13-4	100

Toxicological Data on Ingredients: Carbon Tetrabromide: ORAL (LD50): Acute: 1000 mg/kg [Rat].

# **Section 3: Hazards Identification**

#### Potential Acute Health Effects:

Hazardous in case of eye contact (irritant), of ingestion, of inhalation. Slightly hazardous in case of skin contact (irritant).

#### Potential Chronic Health Effects:

CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Not available. DEVELOPMENTAL TOXICITY: Not available. The substance may be toxic to kidneys, lungs, liver, upper respiratory tract, eyes. Repeated or prolonged exposure to the substance can produce target organs damage.

# **Section 4: First Aid Measures**

#### Eye Contact:

Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Cold water may be used. WARM water MUST be used. Get medical attention.

Skin Contact: Wash with soap and water. Cover the irritated skin with an emollient. Get medical attention if irritation develops.

Serious Skin Contact: Not available.

#### Inhalation:

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

#### Serious Inhalation:

Evacuate the victim to a safe area as soon as possible. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, administer oxygen. If the victim is not breathing, perform mouth-to-mouth resuscitation. Seek medical attention.

#### Ingestion:

Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. If large quantities of this material are swallowed, call a physician immediately. Loosen tight clothing such as a collar, tie, belt or waistband.

Serious Ingestion: Not available.

### Section 5: Fire and Explosion Data

Flammability of the Product: Non-flammable.

Auto-Ignition Temperature: Not applicable.

Flash Points: Not applicable.

Flammable Limits: Not applicable.

Products of Combustion: Not available.

Fire Hazards in Presence of Various Substances: Not applicable.

#### **Explosion Hazards in Presence of Various Substances:**

Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available. Slightly explosive in presence of heat.

Fire Fighting Media and Instructions: Not applicable.

Special Remarks on Fire Hazards: When heated to decomposition it emits toxic fumes of hydrogen bromide.

**Special Remarks on Explosion Hazards:** 

Containers may exploded when heated. Lithium mixed with Carbon Tetrabromide may explode on impact. Explodes on contact with hexacylohexyldilead.

# **Section 6: Accidental Release Measures**

Small Spill: Use appropriate tools to put the spilled solid in a convenient waste disposal container.

Large Spill: Use a shovel to put the material into a convenient waste disposal container.

# Section 7: Handling and Storage

#### Precautions:

Do not ingest. Do not breathe dust. Avoid contact with eyes. Wear suitable protective clothing. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show the container or the label. Keep away from incompatibles such as oxidizing agents.

**Storage:** Keep container tightly closed. Keep container in a cool, well-ventilated area.

# Section 8: Exposure Controls/Personal Protection

#### **Engineering Controls:**

Use process enclosures, local exhaust ventilation, or other engineering controls to keep airborne levels below recommended exposure limits. If user operations generate dust, fume or mist, use ventilation to keep exposure to airborne contaminants below the exposure limit.

### **Personal Protection:**

Splash goggles. Lab coat. Dust respirator. Be sure to use an approved/certified respirator or equivalent. Gloves.

#### Personal Protection in Case of a Large Spill:

Splash goggles. Full suit. Dust respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.

Exposure Limits: Not available.

# **Section 9: Physical and Chemical Properties**

Physical state and appearance: Solid. (Crystals solid.)

Odor: Slight.

Taste: Not available.

Molecular Weight: 331.65 g/mole

**Color:** Colorless. Yellowish-brown.

pH (1% soln/water): Not available.

**Boiling Point:** 189.5°C (373.1°F)

Melting Point: 90.1 (194.2°F)

Critical Temperature: Not available.

Specific Gravity: Not available.

Vapor Pressure: Not applicable.

Vapor Density: Not available.

Volatility: Not available.

Odor Threshold: Not available.

Water/Oil Dist. Coeff.: The product is more soluble in oil; log(oil/water) = 3.4

Ionicity (in Water): Not available.

**Dispersion Properties:** See solubility in water, diethyl ether.

# Solubility:

Soluble in diethyl ether. Very slightly soluble in cold water, hot water. Solubility in water: 0.024 g/100 ml @ 30 deg. C. Soluble in ethanol, chloroform, and other organic solvents. It is very soluble in carbon disulfide.

# Section 10: Stability and Reactivity Data

Stability: The product is stable.

Instability Temperature: Not available.

Conditions of Instability: Incompatible materials

### Incompatibility with various substances: Reactive with oxidizing agents.

Corrosivity: Not available.

Special Remarks on Reactivity: Incompatible with strong oxidizers, lithium, hexacyclohexyldilead.

Special Remarks on Corrosivity: Not available.

Polymerization: Will not occur.

# Section 11: Toxicological Information

Routes of Entry: Inhalation. Ingestion.

Toxicity to Animals: Acute oral toxicity (LD50): 1000 mg/kg [Rat].

Chronic Effects on Humans: May cause damage to the following organs: kidneys, lungs, liver, upper respiratory tract, eyes.

### Other Toxic Effects on Humans:

Hazardous in case of ingestion, of inhalation. Slightly hazardous in case of skin contact (irritant).

Special Remarks on Toxicity to Animals: Not available.

Special Remarks on Chronic Effects on Humans: Not available.

### Special Remarks on other Toxic Effects on Humans:

Acute Potential Health Effects: Skin: May cause skin irritation hyperemia, and moderate edema. Eyes: Causes eye irritant. Severe eye irritant and lachrymator. Can cause corneal injury. Inhalation: Harmful by inhalation. It is irritating to the respiratory tract and causes coughing and wheezing. Acute exposures to high concentrations may cause lung injury, liver injury (fatty liver degeneration, and changes in the liver), kidney injury. High concentrations of vapor may cause central nervous system (CNS) depression. Ingestion: May be harmful if swallowed. May cause diarrhea, chromodacryorrhea, and letharagy. May cause kidney injury (oliguria, aciduria, and hypoosmolality). Chronic Potential Health Effects: Inhalation: Prolonged or repeated inhalation may cause liver injury, bronchitis with cough, phelgm, and or shortness of breath, and lung injury.

# Section 12: Ecological Information

Ecotoxicity: Not available.

BOD5 and COD: Not available.

Products of Biodegradation:

Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.

Toxicity of the Products of Biodegradation: The products of degradation are less toxic than the product itself.

Special Remarks on the Products of Biodegradation: Not available.

# Section 13: Disposal Considerations

#### Waste Disposal:

Waste must be disposed of in accordance with federal, state and local environmental control regulations.

# Section 14: Transport Information

DOT Classification: CLASS 6.1: Poisonous material.

Identification: : Carbon Tetrabromide UNNA: 2516 PG: III

Special Provisions for Transport: Marine Pollutant

# Section 15: Other Regulatory Information

### Federal and State Regulations:

Illinois toxic substances disclosure to employee act: Carbon Tetrabromide Rhode Island RTK hazardous substances: Carbon Tetrabromide Pennsylvania RTK: Carbon Tetrabromide Minnesota: Carbon Tetrabromide Massachusetts RTK: Carbon Tetrabromide New Jersey: Carbon Tetrabromide California Director's List of Hazardous Substances: Carbon Tetrabromide TSCA 8(b) inventory: Carbon Tetrabromide

### **Other Regulations:**

OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200). EINECS: This product is on the European Inventory of Existing Commercial Chemical Substances.

### **Other Classifications:**

#### WHMIS (Canada):

CLASS D-1B: Material causing immediate and serious toxic effects (TOXIC). CLASS D-2B: Material causing other toxic effects (TOXIC).

### DSCL (EEC):

R20/22- Harmful by inhalation and if swallowed. R36- Irritating to eyes. S36/37/39- Wear suitable protective clothing, gloves and eye/face protection. S45- In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible).

#### HMIS (U.S.A.):

Health Hazard: 2

Fire Hazard: 0

Reactivity: 0

**Personal Protection:** E

National Fire Protection Association (U.S.A.):

Health: 2

Flammability: 0

Reactivity: 0

Specific hazard:

#### **Protective Equipment:**

Gloves. Lab coat. Dust respirator. Be sure to use an approved/certified respirator or equivalent. Wear appropriate respirator when ventilation is inadequate. Splash goggles.

# **Section 16: Other Information**

References: Not available.

Other Special Considerations: Not available.

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